# PATENT ABSTRACTS OF JAPAN

(11)Publication number:

2003-313302

(43) Date of publication of application: 06.11.2003

(51)Int.Cl.

CO8G 81/02

(21)Application number: 2002-121733

(71)Applicant: KANEGAFUCHI CHEM IND CO LTD

(22)Date of filing:

24.04.2002

(72)Inventor: HASEGAWA NOBUHIRO

NAKAGAWA YOSHIKI

### (54) CURABLE COMPOSITION

### (57)Abstract:

PROBLEM TO BE SOLVED: To obtain a curable composition which has low viscosity and good workability and of which the cured product has diminution in contamination (including paint staining) by suppressing plasticizer bleeding to the surface of the cured product obtained from the curable composition, has a low modulus and a high elongation, maintains its mechanical properties for a long time and has good adhesion and a high gel fraction.

SOLUTION: The curable composition is composed of a vinyl-based polymer having at least one crosslinkable silyl group and a polyether-based polymer having on the average 1.2 or below of crosslinkable silyl groups.

### **LEGAL STATUS**

[Date of request for examination]

28.02.2005

[Date of sending the examiner's decision of rejection]

[Kind of final disposal of application other than the examiner's decision of rejection or application converted registration

[Date of final disposal for application]

[Patent number]

[Date of registration]

[Number of appeal against examiner's decision of rejection]

[Date of requesting appeal against examiner's decision of rejection]

[Date of extinction of right]

\* NOTICES \*

JPO and NCIP) are not responsible for any damages caused by the use of this translation.

1. This document has been translated by computer. So the translation may not reflect the original

2.\*\*\*\* shows the word which can not be translated.

3.In the drawings, any words are not translated.

#### CLAIMS

[Claim(a)]
[Claim(a)]
[Claim(a)]
[Claim(a)]
[Claim(a)]
[Claim 1] The following two components: The hardenability constituent characterized by containing the virnyl system polymer (I) which has at least one cross-linking silyl radical, and the polyether system polymer (II) which has 1.2 or less cross-linking silyl radicals on the average.
[Claim 2] The hardenability constituent according to claim 1 with which the cross-linking silyl radical of a polyether system polymer (II) is characterized by being in a principal chain end.
[Claim 3] The hardenability constituent according to claim 2 with which the cross-linking silyl radical of a polyether system polymer (II) is characterized by heing only at the one end in a principal chain, and not having at other ends.
[Claim 4] It is a hardenability constituent given in any 1 term among claims 1-3 characterized by containing the vinyl system polymer (I) whose molecular weight distribution is less than 1.8.
[Claim 5] It is a hardenability constituent given in any 1 term among claims 1-4 characterized by containing the vinyl system polymer (I) which is what is manufactured by mainly carrying out the polymerization of the monomer chosen from the group which a principal chain becomes from an acrylic (meta) monomer, an acryloritile system monomer, an aromatic series vinyl system monomer.

Claim 6] The hardenability constituent according to claim 5 characterized by a principal chain containing the vinyl system polymer (I) which is an acrylic (meta) polymer. (Claim 7] The hardenability constituent according to claim 6 characterized by a principal chain containing the vinyl system polymer (I) which is an acrylic polymer. (Claim 8) The hardenability constituent according to claim 7 characterized by a principal chain containing the vinyl system polymer (I) which is an acrylic ester system polymer. (Claim 8) The hardenability constituent according to claim 8 characterized by a principal chain containing the vinyl system polymer (I) which is a butyl acrylate system polymer. (Claim 19] The hardenability constituent according to claim 8 characterized by a principal chain containing the vinyl system polymer (I) which is a butyl acrylate system polymer. (Claim 10] It is a hardenability constituent given in any 1 term to the vinyl system (polymer I) 100 weight section among claims 1-9 characterized by carrying out 5-100 weight section content of the polyether system polymer (III). (Claim 11] Furthermore, it is a hardenability constituent given in any 1 term among claims 1-10 characterized by containing the polyether system polymer (III) which has at least 1.2 or more cross-linking functional groups as the third component. (Claim 12) vinyl — a system — a polymer — (— I —) — 100 — weight — the section — receiving — averaging — 1.2 — a piece — leas than — cross-linking — a lynder — the section — at least — 1.2 — a piece — more than — cross-linking — a functional group — having — a polywether — a system — a polymer (III) — containing — things — the description — \*\* — carrying out — being according to claim 11 — hardenability — a constituent. (Claim 13] it is a hardenability constituent given in any 1 term among claims 1-12 characterized by containing the vinyl system polymer (Wish — a polymer (III) — containing — things — the description — \*\* — carrying out — being according to claim 11 — hardenability — a cons what is manufactured by the living radical polymerization method.

[Claim 14] The hardenability constituent according to claim 13 characterized by a living radical

http://www4.ipdl.ncipi.go.jp/cgi-bin/tran\_web\_cgi\_ejje?u=http%3A%2F%2Fwww4.ipdl.... 2006/10/12

JP,2003-313302,A [CLAIMS] 2/2 ページ

olymerization containing the vinyl system polymer which is an atomic migration radical

polymenzation.
[Claim 15] The hardenability constituent according to claim 14 with which an atomic migration readical polymerization is characterized by containing the vinyl system polymer with which the complex chosen from the transition metal complex which uses the 7th group of the periodic table, eight groups, nine groups, ten groups, or 11 group element as a central metal is made into

a catalyst.

[Claim 18] The hardenability constituent according to claim 15 characterized by containing the vinyl system polymer which is a complex with which the metal complex made into a catalyst is chosen from the group which it becomes from the complex of copper, nickel, a ruthenium, or

iron.

[Claim 17] The hardenability constituent according to claim 16 characterized by containing the viryl system polymer whose metal complex made into a catalyst is a copper complex.

[Translation done.]

http://www4.ipdl.ncipi.go.jp/cgi-bin/tran\_web\_cgi\_ejje?u=http%3A%2F%2Fwww4.ipdl.... 2006/10/12

### \* NOTICES \*

JPO and MCIPI are not responsible for any damages caused by the use of this translation.

- 1. This document has been translated by computer. So the translation may not reflect the original \*\*\*\* shows the word which can not be translated.
- 3.In the drawings, any words are not translated.

### DETAILED DESCRIPTION

## (Detailed Description of the Invention)

[GOUI]
[Field of the Invention] This invention relates to the hardenability constituent containing the virryl system polymer (I) which has at least one cross-linking silyl radical, and the polyether system polymer (II) which has 1.2 or less cross-linking silyl radicals on the average.

[Description of the Prior Art] What has a functional group at a functional group, especially the end with the polymer of the vinyl system obtained by the radical polymerization by one side of the polymer obtained by ionic polymerization or condensation polymerization is hardly put in the polymer obtained by ionic polymerization or condensation polymerization is hardly put in practical use yet. What the acrylic (meta) polymer has the property which is not acquired in polyether system polymers, such as high weatherability and transparency, a hydrocarbon system polymer, or a polyester system polymer, and has an alkenyl radical and a cross-tinking sily! radical in a side chain also in a vinyl system polymer is used for the coating of high weatherability etc. On the other hand, polymerization control of an acrylic polymer is not easy because of the side reaction, and installation of the functional group to an end etc. is dramatically difficult.

fematically difficult.

[0003] If the vinyl system polymer which has an alkenyl radical at the chain end can be obtained by the simple approach, the hardened material which excelled [side chain] in hardened material physical properties as compared with what has a cross-linking radical can be obtained. Therefore, although the manufacturing method has been examined by the researcher of the former many, it is not easy to manufacture them industrially. For example, the synthesis method of the acrybic polymer which has an alkenyl radical st the end (meta) which uses alkenyl radical content disulfide as a chain transfer agent is indicated by JP.1-247403.A and JP.5-255415.A.

[0004] In JP.5-262808.A the vinyl system polymer which has hydroxyl in both ends is compounded using the disulfide which has hydroxyl, and the synthesis method of the acrytic polymer which has an alkenyl radical at the end (meta) is further indicated using the reactivity of hydroxyl.

regroups.

[0005] In JP,5-211922,A, the vinyl system polymer which has hydroxyl in both ends is compounded using the polysulfide which has hydroxyl, and the synthesis method of the acrylic polymer which has a silyl radical at the end (meta) is further indicated using the reactivity of

hydroxyl.

[0008] By these approaches, it is difficult to introduce a functional group into both ends certainly, and the hardened material which has a satisfactory property carnot be obtained. In order to introduce a functional group into both ends certainly, a chain transfer agent must be used for a large quantity, and it is a production process top problem. Moreover, since the usual radical polymerization is used by these approaches, the molecular weight of the polymer obtains and control of molecular weight distribution (ratio of number average molecular weight) are difficult.

[0007] Artificers have invented many to such a Prior art about the vinyl system polymer which has cross-linking functional groups various until now at the end, its manufacturing method, a

has cross-linking functional groups various until now at the end, its manufacturing method, hardenability constituent, and an application (see JP.11-080249,A. JP.11-080250,A. JP.11-

http://www4.ipdl.ncipi.go.ip/cgi-bin/tran\_web\_cgi\_ejje

2006/10/12

JP.2003-313302,A [DETAILED DESCRIPTION]

3/11 ページ

material is a low modulus and high elongation, and rear-spring-supporter maintenance of the machine physical properties is carried out at a long period of time, and it is in offering the constituent which discovers a still better adhesive property, good alkyd paintwork, and a high gel molar fraction. [0014]

feans for Solving the Problem] this invention person etc. completed a header and this invention

the ratio of a monomer to copolymerize in consideration of physical properties, such as oilproof and low temperature-dependencey-characteristics nature.

[0018] Moreover, although definition is not carried out, as for the principal chain of this vinyl system polymer (0), being manufactured by the living radical polymerization is desirable, and it is more desirable that it is an atomic migration radical polymerization. Furthermore, although definition is not carried out, as for an atomic migration radical polymerization, it is desirable to make into a catalyst the complex chosen from the transition metal complex which uses the 1th group of the periodic table, eight groups, nine groups, ten groups, or 11 group element as a central metal, its complex chosen from the group which consists of a complex of copper, nickel, a ruthenium, or iron is more desirable, and especially a copper complex is desirable [a radical polymerization] especially. erization ] especially.

polymerization ] especially. [0019] The location of the cross-linking silyl radical of a viryl system polymer (I) has a desirable end, although definition is not carried out. In addition, although you may have the same functional group as the interior of a principal chain, when asking for rubber elasticity the hardened material made to construct a bridge, it is desirable to have a functional group only at the end. [0020] Although especially the number of the cross-linking functional groups of a vinyl system polymer (I) is not limited, in order to obtain a hardened material with more high cross-linking, it is desirable to have one or more pieces on the average, and it is 1.2 or more pieces [ 3.5 or less ] still more preferably 4.0 or less [ 1.1 or more ] more preferably. [0021] Although especially definition is not carried out, as for the polyether system polymer (II) which averages and has 1.2 or less cross-linking silyl radicals, it is desirable that a cross-linking silyl radicals is in a principal chain end. Moreover, although what has only at the one end in a

005815.A JP.11-116817.A JP.11-116806.A JP.11-080571.A JP.11-080570.A JP.11-130931.A JP.11-100433.A JP.11-116763.A JP.9-272714.A JP.9-272715.A etc.). (0008) For example, the silicon content radical which can construct a bridge by having the hydroxyl group or hydrolysis nature machine combined with the silicon atom, and forming [0008] For example, the silicon content radical which can construct a bridge by having the hydroxyl group or hydroxyls in sture machine combined with the silicon stom, and forming siloxane association with hygroscopic moisture etc. also in a room temperature. The hardened material obtained from the vinyl system polymer which has (it is also hereafter called a "cross-linking silyl radical"), or its constituent Although it excels in thermal resistance or weatherability and especially definition is not carried out, sealing materials, such as a structural clastic sealing compound sealant and a sealing material for multiple glass. Electrical insulation materials, such as electrical and electric equipment and electronic-parts ingredients, such as a solar-battery reser-face sealing agent, and pre-insulation an electric wire, material for cables, A binder, adhesives, clastic achesives, a coating, powder coatings, a coating material, foem, it is available for various applications, such as seals, such as the potting agent for electric electrons, a film, a gashet, a casting ingredient, various modeling materials and wired glass, and rust proofing and the sealing agent for water proof of a glass laminate end face (cutting section), autoparts, and electrical machinery components, various machine parts.

[0009] Also in the aforementioned application, generally the joint and clearance between (various) machinery components, various machine parts.

[0009] Also in the aforementioned application, generally the joint and clearance between (various machinery components, various machine services as physical properties of a hardened material, over a long period of time is very important, being a low modulus, high elongation, and high intensity, and holding those physical properties as physical properties of a hardened material, over a long period of time is very important, being a low modulus, high elongation, and high intensity, and holding those physical properties as physical properties of a hardened material, over a long period

[0010] the hardened material of the hardenability constituent which made it the raw material [UIII] the hardened material of the hardenability constituent which made it the raw material when giart-molecule quantification of the vinyl system polymer was carried out — a low modulus and high elongation — although it can high-intensity-ize, the viscosity of this compound becomes high and the workability at the time of construction worsens. On the contrary, if a vinyl system polymer is hypoviscosity-ized, although workability becomes good, the machine physical properties of a hardened material will fall (a high modulus, low elongation, low-strength-izing). Then, in order to solve this technical problem, a phthalic-acid system plasticizer like the phthalic and substitutions as a functional sense, a solution exhaustive restrict a plasticizer contracts. ester which does not have a functional group, a polyether system plasticizer, etc. are usually [0011]

m(s) to be Solved by the Invention] However, if such a plasticizer is blended so much, in problems, to be solved by the invention Inverver, if such a plasticizer is blended so much, in the hardened material which stiffened the compound, a plasticizer will pass, bleeding (it is also called shift and oil bleeding) will be carried out to a hardened material front face by the time, and problems, such as smeariness, will be produced. Furthermore, the problem of causing lowering of the surface contamination after the contamination to the circumference and paint of a hardened material (sealant etc.), adhesive lowering, the hardness of a hardened material, elongation, etc. by these in also appeared.

the surface contamination and the contamination of a hardened material, elongation, etc. by that is also produced. [0012] Moreover, since the polymer which has the hydrolysis nature silicon radical which two adding-water nature resolvability radicals per silicon atom come to join together was used in many cases, the vinyl system polymer which has such a cross-finking silyl radical take [ when you need the very quick cure rates in the case of using it at the application etc. and low temperature of adhesives etc. especially, the cure rate is not enough, and ] out the flexibility after hardening Crossifinking density needed to be reduced, therefore since crosslinking density was not enough, there was a problem that there was stickiness (surface tuck). [0013] Then, the vinyl system polymer with which this invention has at least one cross-linking silyl radical, And it is the hardenability constituent which uses as a principal component the polyether system polymer which has 1.2 or less cross-finking silyl radicals on the average. Have good workshifty by hypoviscosity and contamination (costing stain resistance is included) of a hardened material is reduced by controlling the bleeding of the plasticizer to the front face of the hardened material which stiffened the hardenability constituent. Moreover, the hardened

http://www4.ipdl.ncipi.go.jp/cgi-bin/tran\_web\_cgi\_ejje

2006/10/12

JP.2003-313302,A [DETAILED DESCRIPTION]

principal chain, and it does not have at other ends is desirable as for the cross-linking silyl radical of this polyether system polymer (II), it averages, and it will not be limited especially

[0022] It is desirable that a vinyl system polymer (I), on the other hand, contains the vinyl system polymer which is the cross-linking silyl radical whose a is 3 among the cross-linking silyl radicals expressed with a general formula (1) in order to raise the cure rate and crosslinking density of a compound.

— SYAR3—a...(1)

- SiYaR3-a ... (1)
however, the inside R of a formula — the alkyl group of carbon numbers 1-20, and the aryl group of carbon numbers 6-20 — The aralkyl radical or (R/) 3SiO of carbon numbers 7-20 - (R' is the univalent hydrocarbon group of carbon numbers 1-20) three R' — being the same — differing — \*\*\*\* — when the Tori ORGANO siloxy radical shown is shown and two or more R exists, they may be the same and may differ. When Y shows a hydroxyl group or a hydrohysis nature machine and two or more Y exists, they may be the same and may differ. a shows 1, 2, or 3.
Moreover, as mentioned above, the location of the cross-linking silyl radical expressed with this formula (1) has the desirable principal chain end of a vinyl system polymer (I), although definition is not carried out.

odiment of the Invention] This invention relates to a hardenability constituent. Furthern

Eimbodiment of the Invention] This invention relates to a hardenability constituent. Furthermore, it is related with the hardenability constituent which contains in detail the viryl system polymer (I) which has following at least one 2 component.cross-linking functional group, and the polyether system polymer (II) which has 1.2 or less cross-linking silyl radicals on the average. Below, the hardenability constituent of this invention is explained in full detail.

(viryl system polymer) > (principal chain) this invention persons The viryl system polymer which has cross-linking functional groups various until now at the polymer end. The manufacturing method. A hardenability constituent, And it is related with an application. Much invention The line came. JP,11-080249,A. JP,11-080250,A. JP,11-05815,A. JP,11-116817,A. JP,11-116806,A. JP,11-080571,A. JP,11-080271,A. JP,11-130931,A. and JP,11-100433,A. — Refer to JP,11-11673,A. JP,9-272714,A. JP,9-272715,A, etc. Although not limited especially as a viryl system polymer (I) of this invention, all the polymers indicated by invention illustrated above can be used suitably.

a vinyl system polymer (I) of this invention, all the polymers indicated by invention illustrated above can be used suitably.

[0024] It is not limited especially as a vinyl system monomer which constitutes the principal chain of the vinyl system polymer of this invention, but various kinds of things can be used. If it illustrates, an acrylic acid (meta), a methyl acrylate (meta), An ethyl scrylate, acrylic-acid (illustrates, an acrylic acid (meta) isopropyl, (Meta) Acrylic-acid-m-butyl, isobutyl acrylate (meta), Muta) Acrylic-acid (meta) acrylic-acid perfluoro ethyl methyl, an acrylic-acid (meta) acrylic-acid acrylic-acid perfluoro methyl acrylic-acid perfluoro ethyl acrylic-acid perfluoro ethyl acrylic-acid perfluoro methyl-acrylic-acid perfluoro ethyl (meta) acrylic-acid (meta) acrylic-acid acrylic-acid acrylic-acid perfluoro methyl-acrylic-acid perfluoro ethyl, (Meta) acrylic-acid acrylic-ac

styrene sulfonic acid, and its salt; Perfluoro ethylene, Fluorine content vinyl system monomers, such as a perfluoro propylene and vinylidene fluoride; Vinyltrimetoxysilane. Silicon content viny system monomers, such as vinyltriethoxysilane; A maleic anhydride, Monoalhyl ester and dialkyl ester of a muleic acid and a maleic acid. A fumaric acid, Monoalkyl ester and dialkyl ester of a ester of a maleic acid and a maleic acid: A furnaric acid, Monosity! ester and diality! ester of a furnaric acid; Maleimide, Methyl maleimide, chtyhraslemide, propyl maleimide, butyhraslemide, betyl maleimide, betyl maleimide, docyl maleimide, betyl maleimide, stearyl maleimide, betyl maleimide system monomers, such as phemyl maleimide and cyclohexyl maleimide; Acrylonitrile, Acrylonitrile; Ac

system monomer. nere — manny above-mentioned monomer 70% or more of preferably among the monomeric units which constitute a vinyl system polymer. (0028) Especially, the styrene system monomer from physical properties etc. and (meta) acrylicacid system monomer of a product are desirable. More preferably, it is an acrylic ester monomer and a methacrylic ester monomer, and is an acrylic ester monomer especially preferably. The point that physical properties, such as hypoviscosity of a compound, a low modulus of a hardened material, high elongation, weatherability, and thermal resistance, are required in the application of general \*\*ex\*\*\*\*\*\*\* to a butyl acrylate system monomer is still more desirable. The copolymer mainly concerned with the ethyl acrylate system monomer is still more desirable. The copolymer mainly concerned with the ethyl acrylate on the other hand in the application as which oibproof [. such as an automotive application.] stc. is required is still more desirable. It can also transpose a part of ethyl acrylate to butyl acrylate or to take that low temperature-dependency—characteristics nature, since it has the inclination to be a little inferior to low temperature-dependency—characteristics nature (cold resistance), although the polymer mainly concerned with this ethyl acrylate is excellent in oilproof. However, since it follows on increasing the ratio of butyl acrylate and oilproof [ the / good] is spoiled, as for the ratio, for the application of which oilproof is required, it is desirable to carry out to 40% or less, and it is more desirable to make it to 30 more% or less. Moreover, in order to improve low temperature—dependency—characteristics nature etc., without spoiling oilproof, it is also desirable to use acrylic-acid 2—methoxy ethyl, acrylic—acid 2—ethoxyethyl, etc. by which oxygen was introduced that the alkyl group of a side chain. However, since it is in the inclination for thermal resistance to be inferior with the installation of an alkoxy group which has ether linkage in a side cha 20 ] [0027] In this invention, it is desirable other mo

mers, copolymerization, and that may carry out block copolymerization, fur to eastwate other monomers, copolymerization, and that may carry or, block copolymerization further and these desirable monomers are contained 40% by the weight ratio in these desirable monomers in that case. In addition, an acrylic acid (meta) expresses an acrylic acid and/, or a methacrylic acid with the above-mentioned transcription. (D028) Although especially the molecular weight distribution of the vinyl system polymer of this invention, i.e., the ratio of the weight average molecular weight (Mw) and number average

http://www4.ipdl.ncipi.go.jp/cgi-bin/tran\_web\_cgi\_ejje

2006/10/12

JP,2003-313302,A [DETAILED DESCRIPTION]

[0036] As for the "living radical polymerization method", research is positively made into groups various in recent years as the example, for example A journal OBU American chemical society (J. Am.Chem.Soc.), 1994, 118 volumes, the thing using a cobalt porphyrin complex as shown in 7943 pages, Macro leakage—at—bulb KYURUZU (Macromolecules), 1994, The thing using radical (U. Am.Chem.Soc.), 1994, 118 volumes, the thing using a cobalt porphyrin complex as shown in 7943 pages, Macro leakage=at-bubk fVVIRUZU (Macromolecules), 1994, The thing using radical scavengers, such as 27 volumes and a nitroxide compound as shown in 7228 pages. The atomic migration radical polymerization." Aftor Transfer Radical Polymerizational RTP) etc. which makes an organic halogenide etc. an initiator and makes a transition metal complex a catalyst is raised, [0037] Also in a "living radical polymerization method", an organic halogenide or a halogenation suffonyl compound An initiator. "The atomic migration radical polymerization method" which carries out the polymerization of the vinyl system monomer by making a transition metal complex into a catalyst As the manufacture approach of a vinyl system polymer of in addition to the description of the above-mentioned "living radical polymerization method" having a halogen comparatively advantageous to a functional-group conversion reaction etc. at the end, and having a specific functional group from the degree of freedom of a design of an initiator or a catalyst being large, it is still more desirable. As this atomic migration radical polymerization method. Matyjaszewski et al. [ for example. ], Journal OBU American chemical society (U. Am.Chem.Soc.) 1995, 28 volumes, 5614 pages, macro leakage-at-bub KYURUZU (Macromolecules) 1995, 28 volumes, 7901 pages, Science (Science) 1996, 272 volumes, 1721 pages, JP.9-208616A, JP.8-41117A, etc. will be mentioned in macro leakage-at-bub KYURUZU (Macromolecules) 1995.

[0038] In this invention, although which approach is used among these living radical polymerizations does not have especially constraint, an atomic migration radical polymerizations.

polymerizations does not neve especially constructed below at the detail, the method is desirable. [0039] Although the living radical polymerization is explained below at the detail, the polymerization using 1 of the control radical polymerizations which can be used for manufacture of the vinyl system polymer later explained before that, and a chain transfer agent is explained. Especially as a radical polymerization using a chain transfer agent (telomer), although not limited, the following two approaches are illustrated as an approach of obtaining a vinyl system polymer with the end structure suitable for this invention.

[MAIN] They are the approach of obtaining the polymer of a halogen end, using halogenated

with the end structure suitable for this envertion.

[0040] They are the approach of obtaining the polymer of a halogen end, using halogenated hydrocarbon as shown in JP.4–132706.A as a chain transfer agent, and a method of obtaining the polymer of a hydroxyl-group end, using a hydroxyl-group content mercaptan or a hydroxyl-group content polysuffide as shown in JP.81–271306.A, JP.2594402.B, and JP.54–47782.A etc. as a chain transfer agent.

[0041] Below, a fiving radical polymerization is explained.

[0042] Among those, the approach using radical scavengers, such as a nitroxide compound, is explained first. Generally in this polymerization, a stable nit ROKISHI free radical (= N-O-) is used as a radical capping agent. As such compounds, although definition is not carried out, the used as a radical capping agent. As such compounds, although definition is not carried out, the nit ROKISHI free radical from annular hydroxy amines, such as a 2, 2, 8, and 6-permutation-1-piperidinyloxy radical and a 2, 2, 5, and 5-permutation-1-pyrrolidinyl oxy-radical, is desirable. As a substituent, a with a carbon numbers (, such as a methyl group and an ethyl group.) of four or less alkyl group is suitable. As a concrete nit ROKISHI free radical compound Although definition is not carried out, 2, 2, 6, a 6-tetramethyl-1-piperidinyloxy radical (TEMPO), A 2, 2, 8, and 6-tetramethyl-1-piperidinyloxy radical, 2, 2, 5, and 5-tetramethyl-1-pyrolidinyloxy radical, 1, 1 and 3, and 3-tetramethyl-2-ISOINDORI nil oxy-radical, N, and N-G tert butylamine oxy-radical etc. is mentioned. Instead of a nit ROKISHI free radical, a free radical with a stable galvinoxyl (galvinoxyl) free radical etc. may be used.

uses. [0043] The above-mentioned radical capping agent is used together with a radical generating agent. It is thought that the resultant of a radical capping agent and a radical generating agent serves as a polymerization initiator, and the polymerization of an addition polymerization nature monomer advances. Although especially both concomitant use rate is not limited, 0.1–10 mols of

molecular weight (Mn) which were measured with gel permeation chromatography, (Mw/Mn) are not limited, it is less than 1.8 preferably, is 1.7 or less more preferably, is 1.6 or less still more preferably, is 1.5 or less still more preferably, is 1.4 or less especially preferably, and is 1.3 or less most preferably hereferably. The GPC measurement by this invention, using chloroform as a mobile phase, a polystyrene gel column can perform measurement and number average molecular weight

prisso, a polystyrene gel column can perform measurement and number average molecular weigh etc. can usually be calculated by polystyrene conversion.

[0029] Although especially a limit does not have the number average molecular weight of the winyl system polymer in this invention, when it measures with gel permeation chromatography, the range of 500–1,000,000 is desirable, 1,000–100,000 are more desirable, and 5,000–50,000 are still more desirable.

Although definition:

still more desirable.

Although definition is not carried out, the synthesis method of the viryl system polymer in Caynthesis method of principal chain. This invention has a desirable control radical polymerization, is more desirable, and is desirable. [ of especially an atomic migration radical polymerization.] [ of a living radical polymerization of These are explained below.

A control radical polymerization radical polymerization method can be classified into "the general radical polymerization method" to which copolymerization of the monomer which has a specific functional group, and the viryl system monomer is only carried out, and the "control radical polymerization method" thich can introduce a specific functional group into the location controlled [ only living was as an extent compound a perceival ext. as a polymerization initiator. polymerization method" which can introduce a specific functional group into the location controlled [ end ], using an azo system compound, a peroxide, etc. as a polymerization initiator. [0030] Although "a general radical polymerization method" is a simple approach, since the monomer which has a specific functional group by this approach is not introduced into a probable polymer, when it is going to obtain a polymer with the high rate of organic-functions-riging, it is necessary to use this monomer for a large quantity considerably, and there is a trouble that the rate of a polymer that this specific functional group is not introduced into reverse by little activity becomes large. Moreover, since it is a free radical polymerization, the trouble that only a large polymer with high viscosity is obtained also has molecular weight distribution. [0031] A "control radical polymerization method" can be classified into the "chain transfer agent method" the vinyl system polymer which has a functional group at the end is obtained, and the "fiving radical polymerization method" the polymer of the molecular weight as a design is mostly obtained by growing without a polymerization growth end causing termination reaction etc., by performing a polymerization using the chain transfer agent which has a further specific functional group.

group.

[0032] Although a "chain trensfer agent method" can obtain a polymer with the high rate of organic-functions-izing, the chain trensfer agent which has the specific functional group of a large quantity considerably to an initiator is required for it, and it has a problem on the financial side also including processing. Moreover, like the above-mentioned "general radical polymerization method", since it is a free radical polymerization, the trouble that it is large and only a polymer with high viscosity is obtained also has molecular weight distribution.

[0033] While termination reaction cannot occur easily and the narrow (Mw/Mn is 1.1 to about 1.5) polymer of molecular weight distribution is obtained though it is the radical polymerization made difficult [ control ] since the termination reaction a "living radical polymerization method" has a high rate of polymerization, and according to coupling of radicals etc. tends to occur unlike these polymerization methods, molecular weight is freely controllable with the preparation ratio of a monomer and an initiator.

these polymerization methods, molecular weight is freely controllable with the preparation ratio of a monomer and an initiator.

[0034] Therefore, a "living radical polymerization method" has narrow molecular weight distribution, and is more desirable as the manufacture approach of a vinyl system polymer of having the above-mentioned specific functional group for the monomer which can obtain a polymer with low viscosity upwards and has a specific functional group since [ of a polymer ] it can introduce into the location of arbitration mostly.

(0055) In addition, although living polymerization means the polymerization to which an end always continues having activity and the chain grows in the narrow sense, the pseudo-living polymerization which grows while that by which the end was inactivated, and the activated thing are generally in an equilibrium state is also contained. The definition in this invention is also the

http://www4.ipdl.ncipi.go.jp/cgi-bin/tran\_web\_cgi\_ejje

2006/10/12

JP,2003-313302,A [DETAILED DESCRIPTION]

radical initiators are suitable to radical capping agent 1 Mol. [0044] As a radical generating agent, although various compounds can be used, the peroxide which may generate a radical is desirable under polymerization temperature conditions. As this peroxide, although definition is not carried out, there are allyls perester, such as peroxy peroxide, although definition is not carried out, there are alkyls perester, such as speroxy carbonates, such as disalkyl peroxide, such as descript peroxide, such as benzoyl peroxide and lauroyl peroxide, JIKUMIRU peroxide, and G t-butyl peroxide, diisopropyl peroxi dicarbonate, and bis(4-t-butyl cyclohexyl) peroxi dicarbonate, t-butyl peroxyoctoate, and t-butyl peroxybenzoate, etc. Especially benzoyl peroxide is desirable. Furthermore, radical generating agents, such as a radical generating nature azo compound like azobisisobutyronitril, can also be used instead of

(0045) Macromolecules An alkoxy amine compound as shown in the following figure may be used as an initiator instead of using together a radical capping agent and a radical generating agent as reported by 1995, 28, and P.2993.

[Formula 1]

If what has functional groups, such as a hydroxyl group as it shown in the above figure, is used when using an alkoxy amine compound as an initiator, the polymer which has a functional group at the end will be obtained. If this is used for the approach of this invention, the polymer which

at the end will be obtained. If this is used for the approach of this invention, the polymer which has a functional group at the end will be obtained. [0047] Although polymerization conditions, such as the monomer and solvent which are used by the polymerization using radical scavengers, such as the above-mentioned nitroxide compound, and polymerization temperature, are not limited, they are the same as that of what is used about the atomic migration radical polymerization explained below, and are not cared about. The more desirable atomic migration radical polymerization method as an atomic migration radical polymerization of this invention is explained. [0048] in this atomic migration radical polymerization, an organic halogenide (for example, the carbonyl compound which has a halogen in an alpha position and the compound which has a halogen in the benzylic position) which has reactant high carbon-halogen association, or a halogenation suffonyl compound is used as an initiator, it illustrates concretely — if it becomes — C6H5-CH2X, C6H5-C(H) (X) CH3, and C6H5-C (X) (CH3) — two (however, the inside of the upper chemical formula, C6 H 5 a phenyl group and X chlorine, a bromine, or iodine) bromine, or iodine) -

bromine, or iodine) — RI-C(CH3) (X)-CO two R2, R1-C(H) (X)-C(O) R2, R1-C(CH3) (X)-C (O) R2 (arnong a formula), For R1 and R2, a hydrogen stom or the slkyl group of carbon number 1-20, an aryl group or an arsikyl radical, and X are chlorine, a bromine, or iodine. R1-C6H4-S02X (for the inside of a formula and R1, a hydrogen atom or the slkyl group of carbon numbers 1-20, an aryl group or an aralkyl radical, and X are chlorine, a bro

[0049] The organic halogenide or halogenation sulfonyl compound which has functional groups ther than the functional group which starts a polymerization as an initiator of an atomic migration radical polymerization can also be used. In such a case, the vinyl system polymer which has a functional group at one principal chain end, and has the growth end structure of an atomic migration radical polymerization at the principal chain end of another side is manufactured. As such a functional group, an alkenyl radical, a cross-linking silyl radical, hydroxyl, an epoxy group, the amino group, an amide group, etc. are mentioned.

[0050] What has the structure which it is not limited as an organic halogenide which has an alkenyl radical, for example, is shown in a general formula (2) is illustrated. R4R5C(X)-R6-R7-C(R3)=CH2 (2)

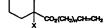
NAHOL(X)-RB-RI-CRI)=CRI (Z)
the inside of a formula, and R3 — hydrogen or a methyl group, and R4 and R5 — hydrogen —
The univalent alkyl group of carbon numbers 1-20, an aryl group, an aralkyl radical or the thing
mutually connected in the other end, and R8 [ or ] – X in which C(0) O- (ester group), —C(0)(keto radical) or o-, m-, p-phenylene group, and R7 may include one or more ether linkage by
direct coupling or the divalent organic radical of carbon numbers 1-20 is chlorine, a bromine, or

iodine.

As an example of substituents R4 and R5, hydrogen, a methyl group, an ethyl group, n-propyl group, an isopropyl group, butyl, a pentyl radical, a hexyl group, etc. are mentioned. R4 and R5 may be connected in the other end, and they may form the arrudar frame.

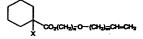
[0051] As an example of an organic histogenide of having the alternyl radical shown by the general formula (2) XCH2C(O) O(CH2) nCH=CH2, H3CC(H) (X) C(O) O(CH2) nCH=CH2, 2(H3C) C(X) C

(0) O(CH2) nCH=CH2, CH3CH2C(H) (X) C(O) O(CH2) nCH=CH2, [0052] [Formula 2]



mentioned caremony, for X, chlorine, a bromine or iodine, and n are the

integer of 0-20) XCH2C (O) O nO (CH2) mCH=CH2, H3CC (CH2) (H) (X) C(O) O(CH2) nO(CH2) mCH=CH2, 2 (H3C) C(X) C(0) O(CH2) nO(CH2) mCH=CH2, CH3CH2C(H) (X) C(0) O(CH2) nO(CH2) nO



(Setting at each above—mentioned ceremony, for chlorine, a bromine or iodine, and n, the integer of 1–20 and m are [ X ] the integer of 0–20) o. m, p~XCH2–C6H4–(CH2) n~CH=CH2, o and m, p~CH3C(H) (X)~C6H4–(CH2) n~CH=CH2, o and m, and p~CH3C — H2C(H) (X)~C6H4–(CH2) n~CH=CH2 (each above—mentioned formula—setting—X — chlorine—J. A bromine or iodine, and n are the integer of 0~20. o. m, p~XCH2~C6H4—n~C (CH2) (CH2) m~CH=CH2, o and m, and p~CH — 3C(H) (X)~C6H4–(CH2) n~C+CH2). occupant of children or iodine, and n in the integer of 1–20 and m are [ X ] the integer of 0~20. o. m, p~XCH2~C6H4—0~(CH2) m~CH=CH2, o and m, and p~CH3C H2—C(H2) m~CH=CH2. occupant of children or iodine, and n, the integer of 1–20 and m are [ X ] the integer of 0~20. o. m, and p~XCH2~C6H4—0~(CH2) m~CH=CH2, o and m, and p~CH3C H2—C(H2) m~CH=CH2 (each above—mentioned formula—setting—X — chlorine, a bromine, or iodine—J, n is the integer of 0~20. o. m, and p~XCH2~C6H4—0~(CH2) m~CH=CH2, o and m, and p~CH3C H2 (each above—mentioned formula—setting—X — chlorine, a bromine, or iodine—J, n is the integer of 0~20. o. m, and p~XCH2~C6H4—0~(CH2) m~CH=CH2, o and m, p~CH3C (CH2) (H)—(X)~C6H4—0~(CH2) m~CH=CH2. or in the integer of 0~20. or in the integer of 0~20 and m are [ X ] the integer of 0~20. The compound further shown by the general formula (3) as an organic halogenide which has an

http://www4.ipdl.ncipi.go.jp/cgi-bin/tran\_web\_cgi\_ejje

2006/10/12

### JP.2003-313302.A [DETAILED DESCRIPTION]

11/11 ページ

-C6H4- 2-O- (CH2) 3Si (CH2) 3, o and m, p-XCH2-C6H4-O- (ОСН3) 3Si (CH2) 3, o and m, p-CH3C (ОСН3) (H) (X) -C6H4-О- 3Si (CH2) 3, o and m, p-CH3CH2C (ОСН3) (H) (X) -C6H4-О-3-Si (CH2) 3, o and m, p-XCH2-C6H4-О- (ОСН3) 2-О- (СН2) 3-Si (СН2) (ОСН3) 3, o and m, p-CH3C(H) (X)-C6H4-O-(CH2)2-O-(CH2)3Si (ОСН3)3, o and m, p-CH3CH2C(H) (X)-C6H4-O-(CH2)2-O-(CH2)3Si (OCH3)3, o and m, p-CH3CH2C(H) (X)-C6H4-O-(CH2)2-O-(CH2)2Si (OCH3)3, o and m, p-CH3CH2C(H) (X)-CH2C(H2)2Si (OCH3)3, o and m, p-CH3CH2C(H2)2Si (OCH3)3, o and m, p-CH3CH2C(H2)2Si (OCH3)2, o and m, p-CH3CH2C(H2)2Si ( (CH2)2-O-(CH2)3Si3 (in each above-mentioned formula) (OCH3), X is chlorine, a broad

### \*\* is mentioned.

The inside of a formula, and R3, R4, R5, R7, R8, R9, R10, a, b, m, X and Y are the same as the

above)
If such a compound is illustrated concretely, 3(CH3O) SiCH2CH2C(H) (X) C6H5, 2(CH3O) (CH3)
SiCH2CH2C(H) (X) C6H5, 3(CH3O) SiCH2)2C(H) (X)—C02R, 2(CH3) SiCH2)2C(H) (X)—C02R, (CH3O) 3SiCH2) 3C(H) (X)—C02R, (CH3O) 3SiCH2) 3C(H) (X)—C02R, (CH3O) 3SiCH2)
4(CH) (X)—C02R, 2(CH3O) (CH3) SiCH2(H) (X)—C02R, (CH3O) 3SiCH2)9C(H) (X)—C6H5, (CH3O) 3SiCH2)9C(H) (X)—C02R, (CH3O) 3SiCH2)9C(H) (X)—C6H5, (CH3O) 3SiCH2)9C(H) (X)—C02R, (CH3O) 3SiCH2)9C(H) (X)—C04R, (CH3O) (

ACC=C(R3)-R7-C(R4XX)-R8-R5 (3)

(the inside of a formula, and R3, R4, R5, R7 and X -- the above -- the same -- R8 -- direct coupling, -C(O) O- (ester group), and - C(O)- (keto radical) or o-, m-, and p-phenylene group

are expressed)

Although R7 is direct coupling or the divisient organic radical (one or more ether linkage may be included) of carbon numbers 1-20, when it is direct coupling, the vinyl group has combined with the carbon which has combined the halogen, and it is an allyl halide ghost. In this case, since carbon-halogen association is activated by the contiguity vinyl group, there may not not necessarily be need of having (CO) 0 set, a phenylene group, etc. as R8, and may be direct coupling. When R7 is not direct coupling, in order to activate carbon-halogen association, as R8, 0 sets, a C(O) C (O) radical, and a phenylene group are desirable.

[0054] If the compound of a general formula (3) is ilbustrated concretely, CH2=CHCH2X, CH2=C(CH3) CH2C, CH2=CHC (X) (CH3) 2, CH2=CH(N) (X) CH3, CH2=CHC (X) (CH3) 2, CH2=CHC(H) (X) CH3, CH3=CHC(H) (X) CH3, CH3=CHC(H) (X) CH3, CH3=CHC(H) (X) CH3, CH3=CHC(CH2) C(H) (X) CH3, CH3=CHCCH) (X)-CO2R, CH3=CHC(CH2) C(H) (X)-CO3R, CH3=CHC(CH2) C(H2) C(H2) C(H3) C(H3)

an be mentioned.

[0055] if the example of a halogenation suffonyl compound of having an alkenyl radical is given, they will be o-, m-, p-CH2=CH-(CH2) n-C5H4-S02X, o-, m-, p-CH2=CH-(CH2) n-O-C6H4-S02X, etc. (setting at each above-mentioned ceremony, for X, chlorine, a bromine or iodine, an n are the integer of 0-20).

[0056] What has the structure which it is not limited especially as an organic halogenide which has the above-mentioned cross-linking silyl radical, for example, is shown in a general formula (4)

is illustrated.

RRFSCKJ-R8-R7-C(H)(R3)CH2-[Si(R9)2-b(Y)b0]m-Si(R10)3-a(Y)a (4)
the inside of a formula, and R3, R4, R5, R6, R7 and X — the above — the same — R9 and R10
All The alkyl group of carbon numbers 1-20, an aryl group, an aralkyl radical, Or (R7) 3SiO ~ (R7 is
the univalent hydrocarbon group of carbon numbers 1-20) three R7—being the same —
differing — \*\*\*\* — when the Tori ORGANO silony radical shown is shown and R9 or two or
more R 10 exist, they may be the same and may differ. When Y shows a hydroxyl group or a
hydrolysis nature machine and two or more Y exists, they may be the same and may differ, a
shows 0, 1, 2, or 3, and b shows 0, 1, or 2, m is the integer of 0-19. However, it shall satisfy that
it is a\*\*mb>=1.

shows 0, 1, 2, or 3, and b shows 0, 1, or 2, m is the integer of 0-19. However, it shall satisfy that it is a+mb)=1.

If the compound of a general formula (4) is illustrated concretely XCH2C (O) 0 nSi (CH2) 3, CCH2 3, (OCH2) 42 (CH3) (X) C (O) 0 nSi (CH2) (CH3) 3, XCH2C(O) 0(CH2) nSi (CH3) (OCH3) 2, CCH23 (CH3) (X) C (O) 0 nSi (CH2) (OCH3) 3, XCH2C(O) 0(CH2) nSi (CH3) (OCH3) 2, CCH3) C (OCH3) (CH3) (CH3)

http://www4.ipdl.ncipi.go.jp/cgi-bin/tran\_web\_cgi\_ejje

2006/10/12